

Faculty of Engineering & Technology

Chemistry 1

Information:

Course Code: CHM 151 Level: Undergraduate Course Hours: 2.00- Hours

Department: Faculty of Engineering & Technology

Instructor Information:

Title	Name	Office hours
Lecturer	Sara Abdel Rsheed Mohamed Mohamed Elhanboushy	
Teaching Assistant	Hagar Magdi Mohamed Mostafa	

Area Of Study:

Overall aims of the course are:

- "Ænrich the students' knowledge about the field of Engineering Chemistry.
- *Ælevate the students' knowledge about the effect of the environment.
- "Ænrich the students with idea about construction materials and cement chemistry.
- ÁDevelop students of practical skills for solving the water pollution problems.

Description:

Gases, Mass balance and heat balance in combustion process of fuels, Solutions, Dynamic equilibrium in physical and chemical processes, Electrochemistry and corrosion, Water treatment, Building materials, Environmental engineering selected chemical industries: fertilizers, dyes, polymers, sugar, petrochemicals, semi-conductors, oil and fats, industrial systems.

Course outcomes:

a. Knowledge and Understanding: :

- 1 Define the main physical and chemical phenomena and terms related to the above-mentioned subjects.
- 2 Describe the general idea about construction of materials and cement chemistry.
- 3 Identify the effect of the environment and problems of water pollution

b.Intellectual Skills::

- 1 Examine different solutions for calculation of numerical problems related to the above-mentioned subjects.
- 2 Analyze chemical reactions and their characteristics to process industries.
- 3 Solve industrial problems in a scientific method.

c.Professional and Practical Skills: :

- 1 Utilize accurate use of different glass wear used for qualitative and quantities chemical analysis.
- 2 Analyze the physical properties of petroleum oil analysis using standard equipment.
- 3 Use the knowledge of Chemistry to solve engineering problems.



d.General and Transferable Skills::

- 1 Collaborate effectively within multidisciplinary team.
- 2 Work coherently and successfully as a part of a team in the Lab and assignments.

Course Topic And Contents :			
Topic	No. of hours	Lecture	Tutorial / Practical
Gasses State	8	4	4
The Properties of Liquids and Solids	8	4	4
Thermo-chemistry	8	4	2
Thermodynamics	4	2	2
Electrochemistry and Corrosion of Metals	4	2	2
Solutions Chemistry	8	4	4
Water and its Treatment	4	2	2
Polymers Chemistry	4	2	2
Chemistry of Cement	4	4	4
Mass balance in combustion process of fuels	8	4	4

Teaching And Learning Methodologies:

Interactive Lecture

Discussion

Problem Solving

Experimental Learning

Cooperative learning

Course Assessment :			
Methods of assessment	Relative weight %	Week No	Assess What
Assignments	5.00		
Final exam	40.00		
Lab Exam	15.00		
Mid-Term Exam	25.00		
Participation and performance	5.00		
Quizzes	10.00		

Course Notes:

Course notes and Handouts

Recommended books:



